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# Polymorphism in binary rare-earth metal germanides. Synthesis, structure and properties of the new hexagonal forms of Tb<sub>3</sub>Ge<sub>5</sub> and Dy<sub>3</sub>Ge<sub>5</sub>

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### 1. Introduction

Rare-earth metal (RE hereafter) germanides are intermetallic compounds, whose structures feature copious bonding patterns [1–3]. This is particularly true for the digermanides, many of which were originally thought to be fully stoichiometric REGe<sub>2</sub> binary phases crystallizing with the ubiquitous AlB<sub>2</sub> and  $\alpha$ -ThSi<sub>2</sub> structure types [3]. However, this rarely holds true and almost exclusively, non-stoichiometric  $REGe_{2-x}$  compounds are formed  $(0 \le x \le 0.5)$ [4-12]. Our recent efforts in exploring this rich structural chemistry have been facilitated through the use of the metal flux method [13] for the synthesis and the crystal growth of several new binary and ternary compounds [4-6,14-18]. On one extreme, we have demonstrated the reactive nature of the metal flux, as evidenced from the formation of the ternary compounds RE<sub>2</sub>InGe<sub>2</sub> (RE=Sm, Gd-Ho, Yb) [14] and  $RESn_{1+x}Ge_{1-x}$  (RE = Y, Gd–Tm) [16]. On the other side, owing to its function as a solvent, the metal-flux has played a beneficial role in our attempts to synthesize the new  $Gd_3Ge_4$  [6] and  $RE_3Ge_5$  compounds (RE = Sm, Gd) [4] or the metastable polymorph of the alkaline-earth phase CaGe<sub>2</sub> [18]. Herein, we exploit again the

#### ABSTRACT

Reported are the synthesis, crystal structure determination and magnetic properties of new polymorphic forms of the rare-earth metal germanides Tb<sub>3</sub>Ge<sub>5</sub> and Dy<sub>3</sub>Ge<sub>5</sub>. Both compounds are isostructural and crystallize with the hexagonal space group  $P\bar{6}2c$  (No. 190, Z=2) with unit-cell parameters a = 6.861(2) Å; c = 8.339(6) Å for Tb<sub>3</sub>Ge<sub>5</sub> and a = 6.8387(10) Å; c = 8.293(2) Å for Dy<sub>3</sub>Ge<sub>5</sub>, respectively. The structures are derivatives of the ubiquitous AlB<sub>2</sub> type and can be regarded as its 6-fold superstructure ( $a' = a \times 3^{1/2}$  and  $c' = c \times 2$ ), arising from the long range ordering of Ge vacancies. They are therefore best described as flat Gelayers, stacked in a hexagonal close-packed manner along the crystallographic *c*-axis, which are separated by layers of rare-earth metal atoms. Magnetic susceptibility measurements reveal that both Tb<sub>3</sub>Ge<sub>5</sub> and Dv<sub>3</sub>Ge<sub>5</sub> exhibit antiferromagnetic order at temperatures below 23 K and 9 K, respectively.

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utility of the flux method for structural studies within a few more *RE*–Ge binary systems and present the synthesis of two new hexagonal polymorphs of the binary compounds Tb<sub>3</sub>Ge<sub>5</sub> and Dy<sub>3</sub>Ge<sub>5</sub>, referred hereafter to as their  $\beta$ -forms (the known orthorhombic polymorphs of Tb<sub>3</sub>Ge<sub>5</sub> [19] and Dy<sub>3</sub>Ge<sub>5</sub> [20] are designated as the  $\alpha$ -forms, respectively). Crystals have been grown from their respective elements using In flux, and their structures elucidated from single-crystal X-ray diffraction. Both compounds crystallize in the hexagonal Sm<sub>3</sub>Ge<sub>5</sub> type [4] with the space group *P*62*c* (Pearson's symbol *hP*16). Along with the structural relationship to the AlB<sub>2</sub> type [3], brought about through a long range ordering of vacant Ge sites, discussed as well are the temperature-dependent magnetization measurements on both compounds and a short description of the structural trends across the series.

#### 2. Experimental

#### 2.1. Synthesis

The reactants were pure elements from Alfa (purity > 99.9% metal basis), which were used as received. They were stored and handled inside an argon-filled glovebox with controlled oxygen and moisture levels below 1 ppm or under vacuum. The reaction conditions were identical to those used for the synthesis of the recently reported  $Sm_3Ge_5$  and  $Gd_3Ge_5$  [4]. Stoichiometric mixtures of the elements in a ratio of *RE*:Ge:In = 3:5:30 were loaded in alumina crucibles (Coors<sup>®</sup>, 2 cm<sup>3</sup>), which were encapsulated in fused silica ampoules. The ampoules were evacuated (*ca.* 10<sup>-3</sup> Torr) and flame-sealed. The reactions took place in muffle furnaces at 1373 K for 3 h, followed by a cooling step (rate 30°/h) to 773 K, where the flux was removed as described in detail previously [14]. Upon opening the sealed tubes, needle-like

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crystals were isolated and later characterized by single-crystal and powder X-ray diffraction as being new hexagonal polymorphs of the known compounds  $Tb_3Ge_5$  [19] and  $Dy_3Ge_5$  [20]. Other metal fluxes were also explored – Cd and Pb for example worked as well – but the quality of the grown crystals was inferior. The crystals of  $Tb_3Ge_5$  and  $Dy_3Ge_5$  exhibit a silver-metallic luster and appear air- and moisture stable over periods of time greater than 1 year.

It was initially found [4], and confirmed by this study, that the cooling rate in the aforementioned flux reactions is of particular importance with regard to the formation of either *RE*<sub>3</sub>Ge<sub>5</sub> polymorph—cooling at slower rates (*e.g.* 5–10°/h) leads to the formation of the orthorhombic phase, while faster cooling (above) yields the hexagonal phase. This could be understood if one realizes that in molten In (or another suitable metal flux), the rates of nucleation and crystallization are under kinetic control and that upon slow cooling, the slower growing  $\alpha$ -*RE*<sub>3</sub>Ge<sub>5</sub> will be the major product; fast cooling will yield the faster growing  $\beta$ -*RE*<sub>3</sub>Ge<sub>5</sub> polymorph as a majority phase. These observations, combined with the fact that polycrystalline samples of  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub> could not be made by arc-melting and annealing are suggestive of  $\beta$ -polymorphs being metastable, kinetic phases, whereas the  $\alpha$ -forms are the thermodynamically stable phases at ambient pressure and temperature. This conclusion is further corroborated by the irreversible single-crystal to single-crystal phase transition, discussed later on, and confirmed by the exclusive formation of the  $\alpha$ -polymorphs by direct fusion of the corresponding elements.

#### 2.2. X-ray diffraction studies

X-ray powder diffraction patterns were taken at room temperature on a Rigaku MiniFlex powder diffractometer using filtered Cu K $\alpha$  radiation. The typical scans were in  $\theta$ - $\theta$  mode ( $2\theta_{max}$  = 80°) with a step-size of 0.05° and 10 s/step counting time. The collected powder patterns were used for phase identification only. This was done using the JADE 6.5 software package [21]. The intensities and the positions of the experimental and the calculated from the crystal structures peaks were in excellent agreement.

Single-crystal data collections were carried out on a Bruker SMART CCD diffractometer using monochromated Mo K $\alpha$  radiation. The crystals were mounted on glass fibers with the aid of Paratone-N oil, which required the use of cryogenic temperatures (120K in this case) in order for the viscous liquid to freeze and prevent the crystals from moving during the data collection. Full spheres of intensity data for crystals from both compounds were collected in 4 batch runs at different  $\omega$  and  $\phi$  angles. Data collections were handled in a routine fashion using the SMART software; the collected frames were integrated using the SAINTplus program [22]. The latter was also used for global unit cell refinement taking into account all reflections. Semi-empirical absorption correction based on equivalents was applied with SADABS [23]. The structures were refined on F<sup>2</sup> using the SHELX package; the coordinates from the isostructural Sm<sub>3</sub>Ge<sub>5</sub> [4] were used as the starting model. In the last refinement cycles, all atomic positions were refined with anisotropic displacement parameters, leading to quick convergences and excellent goodness of fit. Final residuals and other relevant data collection and structural refinement parameters are summarized in Table 1. Positional and equivalent displacement parameters are given in Table 2 along with selected interatomic distances in Table 3. The crystallographic information files (CIF) have also been deposited with Fachinformationszentrum Karlsruhe [76344 Eggenstein, Leopoldshafen, Germany; fax: +49 7247 808 666; email: crysdata@fiz.karlsruhe.de; depository numbers CSD-419457  $(\beta$ -Tb<sub>3</sub>Ge<sub>5</sub>) and CSD-419458  $(\beta$ -Dy<sub>3</sub>Ge<sub>5</sub>)]. Since single crystals of the orthorhombic  $\alpha$ -Tb<sub>3</sub>Ge<sub>5</sub> polymorph were available for first a time, and since its structure had been established before from powder diffraction work only, we carried out structure refinements based on single-crystal X ray diffraction data The results are not discussed herein, but are in excellent agreement with the earlier Rietveld refinement on the orthorhombic α-Tb<sub>3</sub>Ge<sub>5</sub> structure [19] The corresponding CIF has been deposited under CSD-419459.

#### 2.3. Magnetic measurements

Field-cooled dc magnetization (*M*) measurements were completed for both compounds using a Quantum Design MPMS SQUID magnetometer. The measurements were taken on single-crystalline samples in the temperature range from 2 K to 350 K and in an applied field (*H*) of 1000 Oe. The raw magnetization data were corrected for the holder contribution and subsequently converted to molar susceptibility ( $\chi_m = M/H$ ).

#### 3. Results and discussion

#### 3.1. Crystal structure

Only a brief account of the structure of the new hexagonal polymorphs of  $Tb_3Ge_5$  and  $Dy_3Ge_5$  (Pearson's symbol *hP*16) will be given here; for a more comprehensive description and comparison between the structures of the two polymorphs, we refer the reader to the report on similar dimorphism in  $Sm_3Ge_5$  [4].

#### Table 1

Selected single-crystal collection and refinement parameters for  $Tb_3Ge_5$  and  $Dy_3Ge_5.$ 

Tb <sub>3</sub> Ge <sub>5</sub>	Dy <sub>3</sub> Ge <sub>5</sub>	
839.71	850.45	
P62c, 2		
Μο Κα, 0.71073 Α		
120 K		
6.861(2)	6.8387(10)	
8.339(6)	8.293(2)	
340.0(3)	335.90(12)	
$40\mu m \times 40\mu m \times$	$50 \mu\text{m} \times 40 \mu\text{m} \times 40 \mu\text{m}$	
30 µm		
8.203	8.408	
524.16	548.38	
262/18	261/18	
Semi-empirica	l, based on equivalents	
R1 = 0.0193	<i>R</i> 1 = 0.0195	
wR2 = 0.0388	wR2 = 0.0469	
R1 = 0.0258	<i>R</i> 1 = 0.0265	
wR2 = 0.0419	wR2 = 0.0510	
0.89/-1.20e <sup>-</sup> /Å <sup>3</sup>	0.80/-1.29e <sup>-</sup> /Å <sup>3</sup>	
	Tb <sub>3</sub> Ge <sub>5</sub> 839.71 Mo K 6.861(2) 8.339(6) 340.0(3) $40 \mu\text{m} \times 40 \mu\text{m} \times$ $30 \mu\text{m}$ 8.203 524.16 262/18 Semi-empirical R1 = 0.0193 wR2 = 0.0388 R1 = 0.0258 wR2 = 0.0419 $0.89/-1.20e^{-/\tilde{A}^3}$	

<sup>a</sup>  $R1 = \sum ||F_0| - |F_c|| / \sum |F_0|.$ 

<sup>b</sup>  $wR2 = \left[\sum_{v \in V_0} [w(F_0^2 - F_c^2)^2] / \sum_{v \in V_0} [w(F_0^2)^2]\right]^{1/2}$ , where  $w = 1/[\sigma^2 F_0^2 + (A \cdot P)^2 + B \cdot P]$ ,  $P = (F_0^2 + 2F_c^2)/3$ ; A and B: weight coefficients.

A schematic representation of the structure of  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub> is shown in Fig. 1. Using the already communicated ideas [4], this arrangement can be regarded as a long-range ordered superstructure of the non-stoichiometric  $TbGe_{2-x}$  and  $DyGe_{2-x}$ phases  $(x \approx 1/3)$  [3]. The relationship between the defect AlB<sub>2</sub> type and the structure in question is straightforward-the periodic array of planar Ge atoms shown in Fig. 2 represents the topology of the  $\frac{2}{\infty}$  [Ge<sub>5</sub>]-layers in  $\beta$ -*RE*<sub>3</sub>Ge<sub>5</sub>, where every 6th atom from the honeycomb-like arrangement in the idealized  $RE_3Ge_6$ (AlB<sub>2</sub> type, formula tripled for convenience) is vacated in a regular fashion. The resultant super-structure, a', will be related to the unit cell of the sub-structure via the simple geometric relationship  $a' \approx a \times 3^{1/2}$ . In addition, since the newly formed layers are not eclipsed as in the AlB<sub>2</sub>-structure, but rather staggered, the *c*-edge doubles ( $c' \approx c \times 2$ ), yielding an overall 6-fold increase of the cell-volume. Similar approach can be used towards rationalizing the structure of the  $\alpha$ -polymorphs (Pearson's code *oF*64), which in turn can be derived from the  $\alpha$ -ThSi<sub>2</sub> type through ordering of Ge vacancies. In this case, the defect Ge-network can be seen as an array of orthogonal zig-zag chains,  $\frac{1}{\infty}$  [Ge<sub>2</sub>], which are interconnected. The imaginary removal of every 6th atom from the chains can result in a three-dimensional lattice (a', b', c')related to its tetragonal "parent" via the relationships:  $a' \approx a \times 2^{1/2}$ ;

Table 2

Atomic coordinates, isotropic displacement parameters ( $U_{eq}^a$ ) for Tb<sub>3</sub>Ge<sub>5</sub> and Dy<sub>3</sub>Ge<sub>5</sub>.

-					
Atom	Wyckoff index	x	у	Ζ	$U_{\rm eq}({\rm \AA}^2)$
		Tb₃Ge₅			
Tb1	6g	0.3314(1)	0	0	0.0085(2)
Ge1	6h	0.4007(3)	0.3317(3)	1/4	0.0109(3)
Ge2	2d	1/3	2/3	1/4	0.0095(6)
Ge3	2 <i>b</i>	0	0	1/4	0.0104(6)
		Dy <sub>3</sub> Ge <sub>5</sub>			
Dy1	6g	0.33144(9)	0	0	0.0073(2)
Ge1	6h	0.4015(3)	0.3314(3)	1/4	0.0124(4)
Ge2	2d	1/3	2/3	1/4	0.0098(6)
Ge3	2 <i>b</i>	0	0	1/4	0.0111(6)

#### **Table 3** Selected interatomic distances (Å) in Tb<sub>3</sub>Ge<sub>5</sub> and Dv<sub>3</sub>Ge<sub>5</sub>.

Atom pair		Distance				
Tb <sub>3</sub> Ge <sub>5</sub>						
Ge1-	Ge3	2.545(2)				
	Ge2	2.561(2)				
	$Tb \times 2$	2.945(2)				
	$Tb \times 2$	2.970(2)				
	Tb  imes 2	3.450(2)				
Ge2-	$Ge1 \times 3$	2.561(2)				
	Tb  imes 6	3.100(5)				
Ge3-	$Ge1 \times 3$	2.546(2)				
	Tb  imes 6	3.085(2)				
Tb-	$Ge1 \times 2$	2.945(2)				
	$Ge1 \times 2$	2.970(2)				
	$Ge3 \times 2$	3.085(2)				
	$Ge2 \times 2$	3.100(5)				
	Ge1  imes 2	3.450(2)				
	Dy <sub>3</sub> Ge <sub>5</sub>					
Ge1-	Ge3	2.540(2)				
	Ge2	2.558(2)				
	$Dy \times 2$	2.929(1)				
	$Dy \times 2$	2.956(2)				
	Dy  imes 2	3.441(2)				
Ge2-	Ge1 × 3	2.558(2)				
	Dy  imes 6	3.086(2)				
Ge3–	$Ge1 \times 3$	2.540(2)				
	$Dy \times 6$	3.072(2)				
Dy	Ge1  imes 2	2.929(1)				
	$Ge1 \times 2$	2.956(2)				
	$Ge3 \times 2$	3.072(2)				
	$Ge2 \times 2$	3.086(2)				
	$Ge1 \times 2$	3.441(2)				

 $b' \approx a \times 2^{1/2} \times 3$ ;  $c' \approx c$ . More detailed description can be found elsewhere [4,9,10].

From a different standpoint, both "3–5" structures can also be envisioned as built from triangular prisms made of rare-earth metal atoms that share common faces, and which are centered by germanium atoms (Fig. 1 depicts this for  $\beta$ -*RE*<sub>3</sub>Ge<sub>5</sub>). Of worthwhile mention is the fact that not all of these prisms are filled, in fact one out of every six is empty, but this does not cause any distortions in the rare-earth metal sub-lattice.



**Fig. 1.** (a) Crystal structure of the hexagonal  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub>, viewed down the [110] plane and with the unit cell outlined. The rare-earth metal atoms are shown as white spheres while Ge atoms are depicted as smaller, black spheres. The 12-membered polygons of Ge, which are the result of the long-range order of vacant Ge sites are shaded. The trigonal prisms formed by the rare-earth metal atoms are drawn with dotted lines.



**Fig.2.** Schematic representation of the long-range vacancy order in the honeycomblike layers of Ge in the hexagonal  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub>. The corresponding subcell-supercell interrelation is also illustrated by dotted and solid lines, respectively.

As discussed in an earlier study [4], there are several key indications, which suggest that the chemical bonding in the nonstoichiometric TbGe<sub>2-x</sub> and DyGe<sub>2-x</sub> phases ( $x \approx 1/3$ ) cannot be accurately explained using the AlB<sub>2</sub>-type as a model. The first and most significant one has something to do with the fact that the Ge atoms are in special positions in the  $AlB_2$ -type (Wyckoff index 2d), and have no variable parameters [3]. Thus, the Ge–Ge distances will be interrelated by the lattice constants only, which if the published lattice constants for TbGe<sub>2-x</sub> and DyGe<sub>2-x</sub> are taken into consideration [3], amounts to unrealistically short Ge-Ge contacts (on the order of 2.2–2.3 Å). In the 6-fold superstructure (Fig. 2), due to the small distortion around the vacant site in order to compensate for the empty space, the Ge-Ge distances become normal and measure from 2.545(2)Å to 2.561(2)Å in β-Tb<sub>3</sub>Ge<sub>5</sub> and 2.540(2)Å to 2.558(2) Å in  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub> (Table 3), along with the Ge–Ge–Ge bonds angles decreasing from 120° to ca. 103°. Although these parameters are slightly different than those found in elemental Ge [24], they are comparable to the ones in the recently reported binary germanides such as  $RE_3Ge_4$  (RE = Gd through Tm) [6],  $RE_3Ge_5$  (RE = Sm, Gd) [4], EuGe<sub>2</sub> [17] and CaGe<sub>2</sub> [18]. Ultimately, the RE-Ge contacts (in which the rare-earth metal is surrounded by 10 next nearest germanium neighbors) are comparable to those found in the latter compounds and measure between ca. 2.92 Å and 3.45 Å. A comparison of the RE–Ge distances for the known RE<sub>3</sub>Ge<sub>5</sub> compounds with this structure shows a good correlation with the decreasing size of the rare-earth metal cations when moving across the lanthanide series. The Ge-Ge distances, however, are invariant upon changing the RE metal.

#### 3.2. Phase relationships and structural transformations

It has been known for some time that the binary germanides around 50–67 at.% Ge show large stoichiometry breadths and are often described as  $REGe_{2-x}$  ( $0 \le x \le 0.5$ ) [3,9,10]. For the early rareearth metals, the prevailing structure type is the tetragonal  $\alpha$ -ThSi<sub>2</sub> (Pearson's symbol *t*/12), while for the mid-to-late rare-earth metals, the layered hexagonal AlB<sub>2</sub> type (Pearson's symbol *h*P3) is more common [1–3]. We have already examined several such  $REGe_{2-x}$ phases ( $x \approx 1/4$ ,  $x \approx 1/3$ , and  $x \approx 1/5$ ), all of which show evidence for superstructures derived from either  $\alpha$ -ThSi<sub>2</sub> or AlB<sub>2</sub> types through ordering of vacant Ge sites [4–6]. Some of these "line compounds" are not indicated in the corresponding phase diagrams [25], hinting 536

at the possibility that their interpretation may not be as trivial as it may seem at a first glance. Even more surprising is the fact that polymorphic forms and phase transitions are also absent or misrepresented in the diagrams. Take for example the Tb-Ge system [25]—it indicates the possible existence of polymorphic TbGe<sub>2-x</sub> binaries with compositions TbGe<sub>1.6-1.7</sub> (62-63 at.% Ge) and TbGe<sub>1.5</sub> (60 at.% Ge). The structures of the low-temperature forms (LT) remain elusive; the high-temperature form (HT) of TbGe<sub>1.6-1.7</sub> is with the  $\alpha$ -ThSi<sub>2</sub> and the latter with the AlB<sub>2</sub> type. The Dy–Ge system [25], in turn, suggests the possible existence of three polymorphic forms of DyGe<sub>1.5</sub> (type AlB<sub>2</sub> for LT) and only one DyGe<sub>1.63</sub> (structure unknown). It disagrees with a 1966 paper [26], which describes, perhaps erroneously, two  $DyGe_{2-x}$  polymorphs with the  $\alpha$ -ThSi<sub>2</sub> and AlB<sub>2</sub> type. Several structural studies have subsequently shown that the most thermodynamically stable phases in both diagrams at around 63 at.% Ge are indeed Tb<sub>3</sub>Ge<sub>5</sub> and Dv<sub>3</sub>Ge<sub>5</sub> (ordered  $\alpha$ -ThSi<sub>2</sub> superstructures, crystallizing in the *Fdd*<sub>2</sub> space group) [19,20]. These compounds have been prepared by direct fusion of the elements and subsequent annealing at between 800 °C and 1000 °C. Such findings are corroborated by our results using the flux growth technique-the formation of the orthorhombic polymorph is facilitated by slow cooling and/or annealing at intermediate temperatures (500–700 °C), whereas the hexagonal form, which is believed to be the metastable (kinetic) phase can be "trapped" when using fast cooling rates only. Another, even more definitive evidence for the  $\alpha$ - $\beta$  phase relationship is the fact that single crystals of the hexagonal phase ( $\beta$ ), when annealed in vacuum at temperatures between 500 °C and 700 °C for one week, undergo single-crystal to single-crystal phase transition to the orthorhombic form. This transformation was confirmed by X-ray diffraction (see Section 2) and it appears to be irreversible-annealing of the orthorhombic phase in a large temperature range (from 320 °C to 1000 °C), as detailed in previous studies, does not result in any structural change. The results are fully consistent with the  $\alpha$ - $\beta$  phase relationships reported for Dy<sub>3</sub>Ge<sub>5</sub> [20].

#### 3.3. Magnetic properties

Plots of the magnetic susceptibility  $\chi = M/H$  as a function of the temperature for both  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub> are shown in Fig. 3. As seen from the figure, the two compounds exhibit paramagnetic behavior in wide temperature intervals, according to the Curie–Weiss law:  $\chi(T) = C/(T - \theta_p)$ , where  $C = NA\mu_{eff}^2/3k_B$  is the Curie constant and  $\theta_p$  is the Weiss temperature ( $\mu_{eff}$  is the experimental effective moment (in Bohr-magneton units,  $\mu_{\rm B}$ )  $N_{\rm A}$  is the Avogadro number ( $6.022 \times 10^{23} \text{ mol}^{-1}$ ),  $k_{\text{B}}$  is the Boltzmann's constant  $(1.381 \times 10^{-23} \text{ J K}^{-1})$ . Above *ca*. 50 K, the dependence of the inverse susceptibility with the temperature is linear. Applying a linear fit to  $\chi^{-1}(T)$  yields effective moments of 9.63  $\mu_{\rm B}$  and 10.47  $\mu_{\rm B}$ for Tb<sub>3</sub>Ge<sub>5</sub> and Dy<sub>3</sub>Ge<sub>5</sub>, respectively. These values are in very good agreement with the values calculated for free-ion Tb<sup>3+</sup> and Dy<sup>3+</sup> according to  $\mu_{\text{eff}} = g[J(J+1)]^{1/2}$  [27]. For both compounds, cusp-like features at temperature of 23 K for Tb<sub>3</sub>Ge<sub>5</sub> and 9 K for Dy<sub>3</sub>Ge<sub>5</sub> indicate the onset of long-range magnetic order. The magnetic order is probably not simple. The maxima in the  $\chi(T)$  plots can be thought as the respective Néel temperatures, and the Weiss temperatures  $\theta_p$  for Tb<sub>3</sub>Ge<sub>5</sub> and Dy<sub>3</sub>Ge<sub>5</sub> are both negative: -34 K and -27 K, respectively.

It is worth comparing the magnetic properties of the title compound with those, published for other Tb–Ge and Dy–Ge binaries. For example, TbGe<sub>2</sub> (presumed to be a stoichiometric compound) [28] and Tb<sub>3</sub>Ge<sub>5</sub> (orthorhombic polymorph) [20] are both antiferromagnetic with Néel temperatures  $T_N = 42$  K and 17 K, respectively. DyGe<sub>2</sub> and the orthorhombic Dy<sub>3</sub>Ge<sub>5</sub> also undergo antiferromagnetic ordering at 28 K and 12 K, respectively [20]. The difference in



**Fig. 3.** Temperature dependence of the magnetic susceptibility ( $\chi$ ) plots for  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub>. Inverse magnetic susceptibility  $\chi^{-1}(T)$  plots are shown in the insets.

the  $T_N$  values in both cases could indicate that increasing the concentration of Ge vacancies leads to lowering of the corresponding Néel temperature. Of special mention also is the magnetic order in the Dy–Ge compounds described by Sekizawa [26]. This work reports a HT phase with the  $\alpha$ -ThSi<sub>2</sub> type and a LT form with the AlB<sub>2</sub> type, which have Néel temperatures of 6.5 K and 23 K, respectively. These data are inconsistent with the neutron diffraction data on the orthorhombic  $\alpha$ -Dy<sub>3</sub>Ge<sub>5</sub> (LT form) [20] and the discussed herein results on the hexagonal  $\beta$ -polymorph. Such disagreements with earlier studies are not uncommon in the literature, and are most likely due to unrecognized impurities or the presence of secondary phases. After all, the complexity of the corresponding phase diagrams [25] is such that high-quality samples with desired compositions are very difficult to obtain.

# 4. Conclusion

We have reported the synthesis and single-crystal structures of new polymorphs of Tb<sub>3</sub>Ge<sub>5</sub> and Dy<sub>3</sub>Ge<sub>5</sub>. The crystals have successfully been grown from a flux, emphasizing the benefit of the flux-growth method for discovery and facile preparation of single-crystalline compounds.  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub> crystallize with hexagonal structures, which are derivatives of the AlB<sub>2</sub> type. The results from these studies are consistent with the earlier reports on polymorphism in Sm<sub>3</sub>Ge<sub>5</sub>, which all suggest that the hexagonal forms (AlB<sub>2</sub> superstructures, space group *P*62*c*) are the metastable phases, while the more thermodynamically stable phases in the three *RE*–Ge systems (*RE*=Sm, Tb, Dy) are the orthorhombic *RE*<sub>3</sub>Ge<sub>5</sub> ( $\alpha$ -ThSi<sub>2</sub> superstructures, space group *Fdd*2). Temperature dependent magnetization measurements show that  $\beta$ -Tb<sub>3</sub>Ge<sub>5</sub> and  $\beta$ -Dy<sub>3</sub>Ge<sub>5</sub> order antiferromagnetically below 23 K and 9 K, respectively.

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